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Efficient inverted polymer solar cells

G. Li, C.-W. Chu, V. Shrotriya, J. Huang, and Y. Yang^{a)}

Department of Materials Science and Engineering, University of California at Los Angeles, Los Angeles, California 90095-1595

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We investigate the effect of interfacial buffer layers—vanadium oxide (V_2O_5) and cesium carbonate (Cs_2CO_3)—on the performance of polymer solar cells based on regioregular poly-(3-hexylthiophene) and [6,6]-phenyl C60 butyric acid methyl ester blend. The polarity of solar cells can be controlled by the relative positions of these two interfacial layers. Efficient inverted polymer solar cells were fabricated with the structure of indium tin oxide (ITO)/ Cs_2CO_3 /polymer blend/vanadium oxide (V_2O_5)/aluminum (Al). Short-circuit current of 8.42 mA/cm², open-circuit voltage of 0.56 V, and power conversion efficiency of 2.25% under a AM1.5G 130 mW/cm² condition were achieved. The interfacial layers were also used to fabricate polymer solar cells using ITO and a thin gold (Au) layer as the transparent electrodes. The thickness of V_2O_5 layer (10 nm) makes it an effective protective layer for the active layer so that ITO can be used for both the electrodes, enabling highly efficient transparent polymer solar cells (i.e., polymer solar cells with transparent electrodes). Application of this structure for multiple-stacking polymer solar cells is also discussed. © 2006 American Institute of Physics. [DOI: 10.1063/1.2212270]

Fast improvement in polymer solar cells (PSCs) performance in recent years^{1,2} has distinguished this technology as a promising cost effective alternative to silicon based solar cells. PSCs can be utilized to generate truly clean and renewable energy from sunlight. Recently, our group demonstrated very efficient polymer solar cells with 4.4% power conversion efficiency³ (PCE) in regioregular poly-(3-hexylthiophene) (RR-P3HT) and [6,6]-phenyl C60 butyric acid methyl ester (PCBM) bulk-heterojunction (BHJ) structure.⁴ While the external quantum efficiency of polymer solar cells is approaching that of their inorganic semiconductor counterparts,^{2,3,5} limited absorption in solar spectrum remains a major limitation to achieve high PCE. To solve this problem, it is widely accepted that multiple polymer solar cells in tandem, covering different spectral regions, must be used. However, it is generally not easy to realize multiple polymer films in tandem without destroying the underlying polymer layer(s). An alternative approach is to fabricate efficient transparent polymer solar cells (i.e., polymer solar cells with transparent electrodes) with different spectral responses and stack them together.

Deposition of transparent conducting oxides (TCOs) for minimal light loss normally requires sputtering process, which severely damages the underlying organic layers. A suitable interfacial layer can, however, be used to protect the underlying organic layers. Functional interfacial buffer layers (LiF,⁶ CsF,⁷ and AlO_x) have been intensively studied during the development of organic light-emitting devices (OLEDs). LiF has also been reported to enhance polymer solar cell efficiency.⁹ However, the required small thickness (<2 nm) provides insufficient protection of the underlying organic active materials. It was recently demonstrated that some transition metal oxides [e.g., vanadium oxide (V_2O_5), molybdenum oxide (MoO_3)] can be used to replace poly(ethylenedioxythiophene) doped with poly(styrenesulfonate) (PEDOT:PSS) as the anodic buffer layer in polymer solar cells¹⁰ and as an interfacial layer in organic thin-

film transistors.¹¹ These metal oxides are highly transparent, and the device photocurrent is unaffected even when the oxide thickness is up to 20 nm. In an inverted polymer solar cell structure, these metal oxides can provide sufficient protection to the active organic layer while maintaining good device performance. In this letter, we first demonstrate an enhancement in polymer solar cell efficiency by using cesium carbonate (Cs_2CO_3) interfacial buffer layer at the cathode. We then show that the polarity of solar cells can be reversed by changing the position of V_2O_5 (hole injection) and Cs_2CO_3 (electron injection) interfacial layers, independent of the top and bottom electrodes. An efficient inverted polymer solar cell can be fabricated with the device structure: ITO/ Cs_2CO_3 /P3HT:PCBM/ V_2O_5 /metal. Additionally, a thin Au layer is used as the top transparent electrode to fabricate transparent polymer solar cells.

Conventional device fabrication processes were described before.³ Active layer of ~65 nm was spun-coated from RR-P3HT/PCBM 1:1 wt-ratio^{12,13} solution (20 mg/ml each in dichlorobenzene) at 3500 rpm, followed by annealing at 110 °C for 10 min.¹⁴ V_2O_5 (10 nm), Cs_2CO_3 (1 nm), and LiF (1 nm) were thermally evaporated at the rate of about 0.02 nm/s. Solution process Cs_2CO_3 , was described elsewhere.¹⁷ The schematic of device structure is shown in Fig. 1. Thermo-Oriel 150W solar simulator with AM1.5G filter provides 130 mW/cm² illumination, determined by a NREL calibrated Si-detector (with KG-5 color filter) and spectral mismatch is corrected.

Research on functional interfacial materials is a very active area in the field of organic electronic devices. LiF is an effective cathode interfacial layer for both polymer based LEDs¹⁵ and solar cells.⁹ Cs_2CO_3 is a relatively new interfacial material, first reported for OLED applications by Canon group.¹⁶ Unlike LiF, in an OLED the function of Cs_2CO_3 is insensitive to the metal electrode above it. Our group has demonstrated white polymer light-emitting diode (PLED) with 16 lm/W efficiency using this method.¹⁷ In Fig. 2, the *J-V* curves for four different polymer solar cells are shown, with different interfacial layers at the ITO and Al interfaces.

^{a)}Electronic mail: yangy@ucla.edu

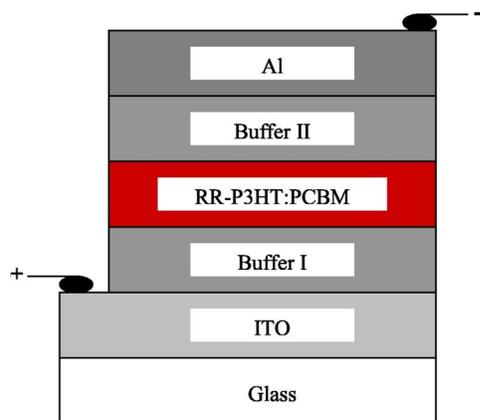


FIG. 1. (Color online) (a) Schematic of the device structure of polymer solar cell fabricated in this study and (b) the molecular structures of active materials in polymer solar cell—RR-P3HT and PCBM.

In the device with no buffer layer (ITO/blend/Al), reasonable photovoltaic effect was observed with short-circuit current density (J_{SC}) of 4.75 mA/cm^2 . However, open-circuit voltage (V_{OC}) and fill factor (FF) are poor at 0.22 V and 28.5% , resulting in PCE of only 0.23% . Modifying the ITO anode by PEDOT:PSS provides significant improvement in the device performance where J_{SC} increases to 7.44 mA/cm^2 , V_{OC} to 0.42 V , FF to 51.8% , and the overall efficiency to 1.25% . Furthermore, insertion of 1 nm thermally evaporated Cs_2CO_3 layer at the polymer/Al interface leads to a reduction in J_{SC} at 5.95 mA/cm^2 . However, the V_{OC} increases to 0.52 V , and an excellent FF of 65.6% is achieved. This results in a PCE of 1.55% , a 25% improvement. These results clearly show that Cs_2CO_3 can act as a functional interfacial layer to enhance polymer solar cell efficiency. The work function of PEDOT:PSS is 5.0 eV , which is 0.3 eV higher than that of ITO (4.7 eV). This work-function increase can explain the increase in V_{OC} by 0.2 eV according to the metal-insulator-metal (MIM) model.¹⁸ This apparently contradicts the common belief that the energy level difference between the donor highest occupied molecular orbital (HOMO) and the acceptor lowest unoccupied molecular orbital (LUMO) levels dominates¹⁹ the V_{OC} in the polymer BHJ solar cell. However, the contact changes from non-Ohmic in the case of ITO to Ohmic for PEDOT:PSS, and both electrodes being Ohmic²⁰ is a necessary condition for the above belief to be valid. Earlier study on Cs_2CO_3 indicated that during thermal

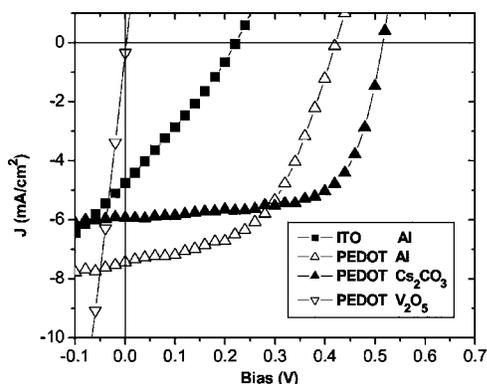


FIG. 2. The effect of interfacial layers on the performance of conventional polymer solar cells. The interfacial layers at ITO/blend and blend/Al interfaces are (a) none, none (solid square); (b) PEDOT:PSS, none (open triangle); (c) PEDOT:PSS, Cs_2CO_3 (solid triangle); and (d) PEDOT:PSS, V_2O_5 (open inverted triangle). The light intensity is 130 mW/cm^2 .

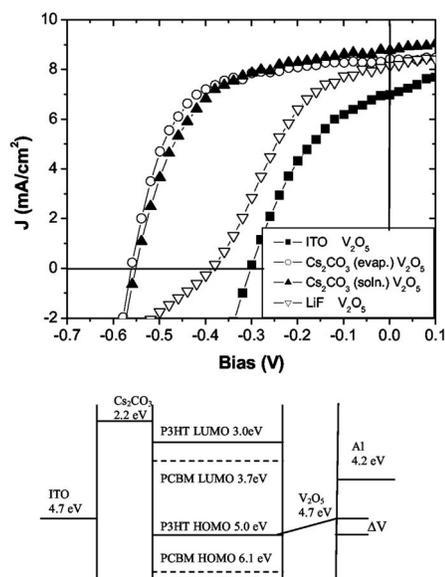


FIG. 3. (a) Current-voltage characteristics of inverted polymer solar cells. The interfacial layers at ITO/blend and blend/Al interfaces are (i) none, V_2O_5 (solid square); (ii) Cs_2CO_3 (evaporated), V_2O_5 (open circle); (iii) Cs_2CO_3 (solution process), V_2O_5 (solid triangle); and (iv) LiF, V_2O_5 (open inverted triangle). (b) Energy level diagrams for various materials in the inverted solar cells.

evaporation, Cs_2CO_3 decomposes into cesium oxide. Depending on the film thickness, the resulting cesium oxide has a field-emission work function of $\sim 1.1 \text{ eV}$ ²¹ because of thermoionic emission. Ultraviolet photoemission spectroscopy (UPS) measurements conducted in our laboratory on thermally evaporated Cs_2CO_3 films show a work function of 2.2 eV .²² The polymer/ Cs_2CO_3 contact is therefore Ohmic. An increase in V_{OC} by only 0.1 V , despite the work-function difference between Cs_2CO_3 and Al of 2 eV , agrees well with the earlier observation by Brabec *et al.*¹⁹ and indicates Fermi-level pinning. The ITO/ V_2O_5 /blend/Al device shows the same polarity as that of ITO/PEDOT:PSS/blend/Al device. The V_{OC} for the former (J - V curve not shown here) is 0.38 eV , also significantly higher compared to bare ITO electrode. The HOMO level of a thermally evaporated V_2O_5 film was determined by UPS to be 4.7 eV which is identical to that of ITO. The most plausible reason for V_{OC} enhancement is the formation of surface dipoles between V_2O_5 and an active layer, which causes an upward shift in work function of at least 0.2 eV . These results indicate that V_2O_5 can be considered as an effective hole injection layer, like PEDOT:PSS, with a similar effective work function. This is further evidenced by the J - V curves for an ITO/PEDOT:PSS/polymer blend/ V_2O_5 /Al device, where no photovoltaic effect was observed (see Fig. 2). The anode contact therefore has an important effect on polymer solar cell performance.

Figure 3 shows J - V curves for various inverted polymer solar cell structures. The ITO/polymer blend/ V_2O_5 (10 nm)/Al inverted solar cell has $J_{SC}=6.97 \text{ mA/cm}^2$, $V_{OC}=0.30 \text{ V}$, $\text{FF}=41.2\%$, and PCE of 0.66% . This provides further evidence for the presence of surface dipoles that enhance the V_2O_5 work function by $\sim 0.3 \text{ eV}$. Based on the results so far, an efficient inverted polymer solar cell can be achieved with the structure ITO/ Cs_2CO_3 /polymer blend/ V_2O_5 /Al, where Cs_2CO_3 was either thermally evaporated or spin coated. The J - V curves in Fig. 3 for solar cells with thermally evaporated (1 nm , open circle) and solution pro-

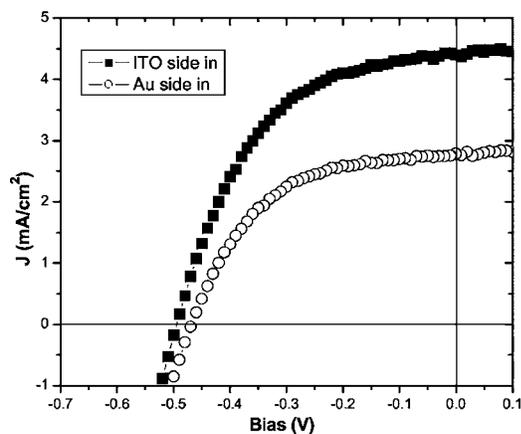


FIG. 4. J - V curves for transparent polymer solar cells illuminated from two transparent electrodes—ITO and Au.

cessed (solid triangle) Cs_2CO_3 clearly demonstrate efficient inverted solar cells. The J_{SC} , V_{OC} , and FF are very similar for the evaporated (8.42 mA/cm^2 , 0.56 V , and 62.1%) and the solution processed (8.78 mA/cm^2 , 0.55 V , and 56.3%) device. The overall efficiencies are 2.25% and 2.10% , respectively. Therefore, inserting V_2O_5 and Cs_2CO_3 interfacial layers can result in efficient conventional as well as inverted polymer solar cells. LiF (1 nm, electron injection layer) and V_2O_5 (10 nm, hole injection layer) were also used to fabricate inverted solar cells. This device has a current density comparable to the device with Cs_2CO_3 and V_2O_5 , but an antidiode behavior results in low V_{OC} (0.39 V), FF (40.7%), and PCE (0.99%). A thin LiF layer has been reported to work well with a wide range of metals (Ca, Al, and Au) in conventional device configuration. The presence of an antidiode in the inverted configuration indicates that the LiF growth pattern (on ITO versus on polymer) might have a significant impact on device performance.

In conventional device structure, introducing 1 nm thermally evaporated Cs_2CO_3 reduces the device photocurrent but improves V_{OC} and FF significantly, indicating possible physical damage. However, in the inverted structure, where Cs_2CO_3 is deposited on ITO substrates, all three parameters are improved.

Based on information collected, we can treat V_2O_5 as a hole injection layer with “effective” work function of $\sim 5.0 \text{ eV}$ and Cs_2CO_3 as an electron injection layer with very low work function, both of which provide Ohmic contacts. The polarity of the device is decided by the relative positions of these two interfacial layers and is insensitive to the conducting electrodes. The energy level diagrams for various inverted configurations are illustrated in Fig. 3(b).

Due to the presence of a thick (10 nm) V_2O_5 layer which protects the underlying polymer, the inverted configuration is especially suitable for making transparent solar cells. We replaced a thick Al top electrode with 12 nm of Au in the inverted structure. The J - V curves for this transparent polymer solar cell, with light incident from ITO and the Au side, are shown in Fig. 4. When illuminated from the ITO side, the device shows an overall efficiency of 0.85% . However, when illuminated from the semitransparent Au electrode, the PCE is 0.52% . The difference between the two J - V curves is due to the partial loss by the reflection and absorption at the semitransparent Au electrode. In a transparent solar cell, the light absorption is less than that in a device with a reflecting

metal electrode. Similar to inorganic solar cells, the V_{OC} of organic solar cells also increases with increasing incident light intensity. The slight reduction in the V_{OC} of transparent solar cells can therefore be explained by the less effective light intensity. When designing a tandem polymer solar cell, the optical losses due to the first transparent solar cell should be reduced. To provide sufficient electrical conductance, Au layer thickness has to be sufficient and the optical loss at Au electrode becomes significant. However, the inverted solar cell structure has a metal oxide layer that is not only transparent but also provides effective protection to the polymer layer. A transparent conductive oxide electrode, such as ITO, can therefore be deposited without compromising device performance. This structure thus provides a very efficient method for realizing tandem polymer solar cells for improving device efficiency.

To summarize, we have fabricated efficient conventional and inverted polymer solar cells by using different functional interfacial layers. Efficiency up to 2.25% has been achieved for an inverted polymer solar cell with FF as high as 62.1% . Preliminary efforts have also demonstrated transparent polymer solar cells with 0.85% efficiency.

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- ¹S. E. Shaheen, C. J. Brabec, N. S. Sariciftci, F. Padinger, T. Fromherz, and J. C. Hummelen, *Appl. Phys. Lett.* **78**, 841 (2001).
- ²F. Padinger, R. S. Rittberger, and N. S. Sariciftci, *Adv. Funct. Mater.* **13**, 85 (2003).
- ³G. Li, V. Shrotriya, J. Huang, Y. Yao, and Y. Yang, *Nat. Mater.* **4**, 864 (2005).
- ⁴N. S. Sariciftci, L. Smilowitz, A. J. Heeger, and F. Wudl, *Science* **258**, 1474 (1992).
- ⁵P. Schilinsky, C. Waldauf, and C. J. Brabec, *Appl. Phys. Lett.* **81**, 3885 (2002).
- ⁶L. S. Hung, C. W. Tang, and M. G. Mason, *Appl. Phys. Lett.* **70**, 152 (1997).
- ⁷G. E. Jabbour, B. Kippelen, N. R. Armstrong, and N. Peyghambarian, *Appl. Phys. Lett.* **73**, 1185 (1998).
- ⁸F. Li, H. Tang, J. Andereg, and J. Shinar, *Appl. Phys. Lett.* **70**, 1233 (1997).
- ⁹C. J. Brabec, S. E. Shaheen, C. Winder, N. S. Sariciftci, and P. Denk, *Appl. Phys. Lett.* **80**, 1288 (2002).
- ¹⁰V. Shrotriya, G. Li, Y. Yao, C.-W. Chu, and Y. Yang, *Appl. Phys. Lett.* **88**, 073508 (2006).
- ¹¹C. W. Chu, C. W. Chen, S. H. Li, E. H. Wu, and Y. Yang, *Appl. Phys. Lett.* **86**, 253503 (2005).
- ¹²V. Shrotriya, J. Ouyang, R. J. Tseng, G. Li, and Y. Yang, *Chem. Phys. Lett.* **411**, 138 (2005).
- ¹³D. Chirvase, J. Parisi, J. C. Hummelen, and V. Dyakonov, *Nanotechnology* **15**, 1317 (2004).
- ¹⁴G. Li, V. Shrotriya, Y. Yao, and Y. Yang, *J. Appl. Phys.* **98**, 043704 (2005).
- ¹⁵T. M. Brown, R. H. Friend, I. S. Millard, D. J. Lacey, J. H. Burroughes, and F. Cacialli, *Appl. Phys. Lett.* **79**, 174 (2001).
- ¹⁶T. Hasegawa, S. Miura, T. Moriyama, T. Kimura, I. Takaya, Y. Osato, and H. Mizutani, *Dig. Tech. Papers—Soc. Inf. Disp. Int. Symp.* **35**, 154 (2004).
- ¹⁷J. Huang, G. Li, E. Wu, Q. Xu, and Y. Yang, *Adv. Mater. (Weinheim, Ger.)* **18**, 144 (2006).
- ¹⁸I. D. Parker, *J. Appl. Phys.* **75**, 1656 (1994).
- ¹⁹C. J. Brabec, A. Cravino, T. Fromherz, N. S. Sariciftci, M. Minse, L. Sanchez, and J. C. Hummelen, *Adv. Funct. Mater.* **11**, 374 (2001).
- ²⁰V. D. Mihailetchi, P. W. M. Blom, J. C. Hummelen, and M. T. Rispen, *Appl. Phys. Lett.* **94**, 6849 (2003).
- ²¹T. R. Briere and A. H. Sommer, *J. Appl. Phys.* **48**, 3547 (1977).
- ²²J. Huang and Z. Xu (unpublished).